

## A Nanodrop Spectrophotometric Method and Stability Indicating for Determination of Amlodipine Besylate in Pharmaceutical Formulations of Kurdistan of Iraq

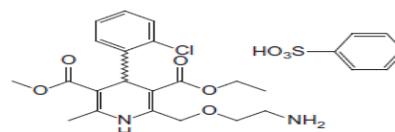
Faroq Omer Qasim<sup>a,\*</sup>, Nidhal Meena Sher Mohammed<sup>b</sup><sup>a</sup> Department of Horticulture, Akre technical college, DPU, Kurdistan Region, Iraq- ([faroq.omer@dpu.edu.krd](mailto:faroq.omer@dpu.edu.krd))<sup>b</sup> Department of Chemistry, Faculty of Science, University of Zakho, Kurdistan Region, Iraq- ([nidhal.mohammed@uoz.edu.krd](mailto:nidhal.mohammed@uoz.edu.krd))**Received: Dec., 2020 / Accepted: Jan., 2021 / Published: Mar., 2021** <https://doi.org/10.25271/sjuoz.2021.9.1.783>**ABSTRACT:**

A nanodrop spectrophotometric method was developed and validated for determination of amlodipine besylate (AB) in bulk and tablet dosage form. The maximum absorption of amlodipine was shown at 357 nm using acetonitrile as a solvent. The developed method was found to be linear ( $R^2 = 0.9990$ ) within the concentration range of 210  $\mu\text{g/mL}$ . The precision study showed acceptable values of RSD% (less than 1%). LOD and LOQ values were found to be 0.34 and 1.14  $\mu\text{g/mL}$ , respectively. Accuracy study showed good recovery 99% Amlodipine and 98.88% Amloneer, in locally commercial tablets. The present method was applied successfully for stability indicating study of AB in Amlodipine and Amloneer products manufactured in (Erbil and Sulaymaniyah, respectively) / Kurdistan of Iraq. The stability-indicating study was investigated under acidic, basic, oxidative, photolytic, and thermal conditions. The results of both products showed that AB is unstable in acidic, alkaline, and oxidative conditions under heating at 60°C up to 5 hrs. While under photolytic and thermal conditions, the degradation percentage was less than 15% indicating to the stability of AB in both Amlodipine and Amloneer tablets according to International Conference on Harmonization (ICH) guideline of drugs. It can be concluded that the main factor that affects the degradation of AB is the passage of time.

**KEYWORDS:** Amlodipine, Nanodrop spectrophotometer, Development, Validation, Stability Indicating.**1. INTRODUCTION**

High pressure or hypertension is a chronic medical condition that raises high pressure in the arteries (Aram *et al.*, 2003). Hypertension or high blood pressure is one of the most common vascular disorders causing a major risk factor for coronary heart disease, stroke, peripheral, cerebrovascular, and peripheral vascular disorders. For treatment of high blood pressure, anti-hypertensive medications are usually used which include different classes of drugs (Khatib *et al.*, 2005).

Amlodipine besylate (AB) is an antihypertensive drug and chemically described as 3-ethyl-5-methyl (4R)-2-[(2-aminoethoxy) methyl] -4-(2-chlorophenyl)-methyl-1,4-dihydropyridine-3,5-dicarboxylate benzenesulfonate with the chemical structure shown in Figure 1 (Abdoh *et al.*, 2004). It belongs to a class of medications called dihydropyridine which is a calcium channel blocker acts by relaxing the smooth muscle in the arterial wall, and causes reducing blood pressure (Bernard *et al.*, 2011). In reviewing the literature, many analytical methods were developed and validated for determination of AB alone or in combination with other drugs including spectrophotometry (Pradeep *et al.*, 2009; Priyanka *et al.*, 2012; Kumar *et al.*, 2019; Kushwaha *et al.*, 2019; Attimarad *et al.*, 2019), thin layer chromatography (TLC) (Meyyanathan *et al.*, 2005; Sharma *et al.*, 2014; Mathew *et al.*, 2014), high performance liquid chromatography (HPLC) Chitlange *et al.*, 2008; (Jadhav *et al.*, 2013; Rima *et al.*, 2013; Mohamed *et al.*, 2019) and kinetic (Hemmateenejad *et al.*, 2009; Mahmoud *et al.*, 2012).

**Figure 1.** Chemical structure of Amlodipine besylate.

Stability indicating is an important part of the process of pharmaceutical product development. The purpose of stability indicating is to provide evidence on how the quality of the drug product varies with time under various environmental effects (Chatuevedi *et al.*, 2004). Forced degradation study involves the possibility of degradation of drug product under extremely conditions such as acid, alkaline, oxidative, thermal, and light (Chakraborty *et al.*, 2018). Study the stability of drug allows selecting proper formulation and packaging for appropriate storage and shelf-life conditions, which are important for regulatory documents of developing any pharmaceutical product. The present study aims to develop and validate a new analytical method using nanodrop spectrophotometer for routine analysis and stability-indicating study of amlodipine besylate in pure form and pharmaceutical tablets.

**2. MATERIALS AND METHODS****2.1 Chemicals**

Pharmaceutically active ingredient (99 % purity) of amlodipine besylate was purchased from Awa medical company in Kurdistan of Iraq. Commercial tablets of Amlodipine (100 mg) and Amloneer (100 mg) from Awa and Pioneer companies in (Erbil and Sulaymaniyah, respectively) Kurdistan of Iraq were procured from the local drug shops. Regents of hydrochloric acid and sodium hydroxide (Scharlau, Spain), hydrogen peroxide (Roth, Germany) were analytical grades. Acetonitrile, methanol, and water (Merck, Germany) were HPLC grade.

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## 2.2 Instrumentation

Spectrophotometric analysis was carried using nanodrop spectrophotometer (2000C Micro volume) from Thermo Scientific. UV lamp (UVC-215 TS 8W, 220-240v, 50/6 Hz) was used for photo degradation study. The analytical balance was Voyager<sup>®</sup>. The water bath shaker was Elmasonic P (100W, 80 KHz) and the oven was Lab Tech (LVO-2030).

## 2.3 Preparation of stock and working solutions

Weighed accurately 100 mg of pure amlodipine besylate (AB) was dissolved in acetonitrile then transferred to 100 mL volumetric flask, and final volume made up to mark with acetonitrile to prepare 1000 µg/mL of stock standard solution. This stock standard solution was kept in a fridge and used to prepare different concentrations of working solutions.

## 2.4 Method Optimization

The maximal absorption wavelength ( $\lambda_{max}$ ) of the AB solution 50 µg/mL was determined using a nanodrop spectrophotometer within a wavelength range of 200-800 nm against acetonitrile as blank,  $\lambda_{max}$  was found to be 357 nm.

**Preliminary** solubility study of AB was also investigated testing different solvents namely water, methanol and acetonitrile.

## 2.5 Method Validation

According to the International Conference on Harmonization (ICH) guideline, the nanodrop spectrophotometric method was validated in terms of system, linearity, precision, LOD, LOQ, accuracy, specificity, and robustness.

## 2.6 Forced degradation study

Stability study of AB in pharmaceutical formulation was performed under Examine conditions including acidic (0.1N HCl), alkaline (0.1N NaOH), oxidant (5% (v/v) H<sub>2</sub>O<sub>2</sub>), thermal (60 °C), and photolytic (sunlight, UV, and dark) at various periods. Two products of Amlodipine and Amloneer, were tested in this study.

**2.6.1 Preparation of stock solution of formulation:** Twenty tablets (2.0 g) from each commercial product, (each tablet of 100 mg contains 5 mg of AB) were weighed accurately and finely powdered. Tablets powder equivalent to 100 mg of AB was taken in a 100 mL beaker and 30 mL of acetonitrile was added to the beaker then sonicated for 15 minutes. The solution was filtered in 100 mL volumetric flask using Whatman filter paper (0.45 µm) and then the volume was made up to the mark with acetonitrile to form a stock solution containing 1000 µg/mL of AB. From this stock solution of formulation, working solutions at different concentration were prepared.

**2.6.2 Preparation of blank solution:** To prepare blank solution, 12.5 mL of each reagent used for degradation (including 0.1 N HCl, 0.1 N NaOH, and 5% H<sub>2</sub>O<sub>2</sub>) was transferred to 25 mL volumetric flask and the volume was completed with acetonitrile and heated at 60°C. From this solution, 2 mL of an aliquot taken in separate 10 mL volumetric flask at different time intervals and neutralized with proper reagent, and then the final volume was made up to mark with acetonitrile and the absorbance was measured at 357 nm.

**2.6.3 Acidic condition:** In 25 mL volumetric flask, 12.5 mL from the stock solution (100 µg/mL) of drug was transferred and 12.5 mL of 0.1 N of HCl was added, and then heated to 60°C. At the beginning 2 mL of this solution was taken at room temperature. Then, another 2 mL from this solution was transferred into separate volumetric flask (10 mL) at different times up to 5 hrs. After neutralizing with a suitable reagent the total volume was completed with acetonitrile to obtain 10 µg/mL of AB. The absorbance was measured against blank at 357 nm.

**2.6.4 Basic condition:** The same procedure in section 2.6.3 was followed replacing 0.1 N of HCl with 0.1N of NaOH.

**2.6.5 Oxidation condition:** The same procedure in section 2.6.3 was followed replacing 0.1 N of HCl with 5% of H<sub>2</sub>O<sub>2</sub>.

**2.6.6 Photo degradation:** Three replicates of 12.5 mL from a stock solution (100 µg/mL) of drug were transferred into 25 mL volumetric flasks to prepare three solutions containing 50 µg/mL of AB. These solutions were exposed to different circumstances including sunlight, UV, and dark up to 6 days. From each solution, 2 mL was transferred into 10 mL of volumetric flask after every 1 day, and the final volume was made up with acetonitrile to form solution containing 10 µg/mL of AB. The absorbance was measured at 357 nm against acetonitrile as blank.

**2.6.7 Thermal degradation:** Accurate weight of 2.0 g (20 tablets) of drug was taken and powdered by mortar and then exposed in an oven at 60 °C up to 6 hrs. From this powdered, 20 mg of exposed drug was dissolved in acetonitrile then transferred into 10 mL volumetric flask that was sonicated for 15 minutes and the final volume was made with acetonitrile to form a solution containing 10 µg/mL of AB. The absorbance was measured (at  $\lambda_{max}$  = 357 nm) against acetonitrile as blank.

## 3. RESULT AND DISCUSSION

### 3.1 Method Optimization

Developing and validating analytical method aims to confirm a suitable method for analysis a particular analyte with specific, accurate, and precise results. The main objective for that is to improve the conditions and parameters followed in the development and validation methods, therefore a wide variety of procedures need to be tested.

**3.1.1 Scanning and determination of  $\lambda_{max}$ :** The spectrum of AB showed maximum absorption at 357 nm as shown in Figure 2. This wavelength was used later for all study measurements.

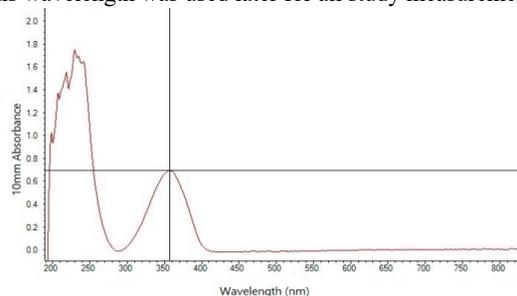


Figure 2: UV-Visible spectrum of AB for 50 µg/mL

**3.1.2 Type of solvent:** The effect of the solvent was studied testing different solvents namely water, methanol and acetonitrile. 1 mg of pure AB was weighed and solubility was checked in 10 mL solvent. The drug was found to be freely soluble in acetonitrile and methanol than water. Measuring the absorbance showed that using acetonitrile gave the highest absorbance (0.180) at 357 nm than in water (0.167) and, methanol (0.134). Therefore, acetonitrile was used as a solvent in this study.

### 3.2 Method Validation

Method validation is process for the collection and evaluation of data to establish precise evidence that the analytical method is capable of producing quality products (Sekhar et al., 2020). The developed method was validated according to ICH guidelines for the linearity, precision, accuracy, specificity, robustness, LOD and LOQ.

**3.2.1 Linearity:** The linearity of the analytical method is the ability of the method to produce test results that are directly proportional to the analyte concentration in the sample (within a given range)(Prajakta et al., 2018). The linearity was investigated at concentration range of 2-10 µg/mL. The linear relationship expressed by the coefficient of determination (R<sup>2</sup>) of the regression line which was found to be 0.9990 showing a good linearity. The ICH acceptance criteria of the linear regression is ≥ 0.9900 (Ferenczi-Fodor et al., 2001).

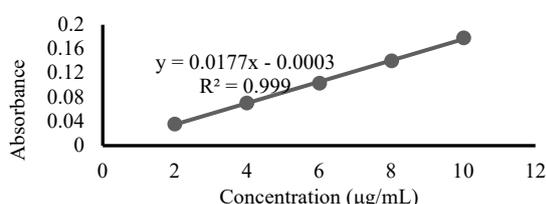


Figure 3: Calibration curve of AB

**3.2.2 Precision:** The precision was evaluated by repeated injection, intra-day and inter-day study of 5 µg/mL of AB solution. Table 1 shows RSD% values which are less than 1% indicating to good precision according to standard criteria of ICH.

Table 1: Evaluation of precision study

Type of precision	RSD%
Repeatability (n=5)	0.79
Intraday (n=3)	0.79
Interday (n=3)	0.70

**3.2.3 LOD and LOQ:** Limit of detection (LOD) of an analytical technique is the lowest concentration that can be detected and distinguished from zero but may not always be calculated as an exact measure. Whereas, limit of quantification (LOQ) is the lowest concentration of the analyte in a sample that can be quantitatively determined with suitable precision and accuracy ( Prajakta et al., 2018). LOD and LOQ can be calculated from the following formulas:

$$LOD = 3.3 * \sigma / S \quad (1)$$

$$LOQ = 10 * \sigma / S \quad (2)$$

Where  $\sigma$  = standard deviation of response, S = slope of the calibration curve, Table 2, shows LOD and LOQ values at minimum levels of amlodipine.

Table 2: LOD and LOQ values obtained from the calibration curve of concentrations (2-10 µg/ml) of AB

Parameter	Value
Linearity Range (µg/mL)	2-10
Regression equation	$y = 0.0177x - 0.0003$
R <sup>2</sup>	0.9990
Slope	0.0177
Intercept	- 0.0003
SD	0.002
LOD µg/mL	0.34
LOQ µg/mL	1.14

**3.2.4 Recovery:** The validity and accuracy of the proposed method were assessed by recovery study testing three replicates sample at concentration of 10 µg/mL of each drug (Avalodipine and Amloneer). The recovery results were 99.00% for Avalodipine and 98.88% for Amloneer.

**3.2.5 Specificity:** Specificity is the ability to assess the analyte in the presence of components that may be expected to be present. Typically, these might include impurities and matrix ( Nagamani et al., 2013). The specificity of the present method was determined by adding (starch and magnesium) to the standard solution of AB and calculating the percent of the recovery for three replications which was found to be 99.8% with RSD% 0.32. The resultant showed that there was not interference from excipients in the analysis of standard AB.

**3.2.6 Robustness:** Robustness is measure of capability of method to remain unaffected by small variations in method parameters ( Koumudi et al., 2012). The effect of change in the wavelength from 355 to 359, in the absorbance was studied. Standard solution of AB (5 µg/mL) was prepared and analyzed at different wavelength. The RSD% of absorbance measurement was found to be less than 1 %.

Validation of the present analytical method showed it is linear, precise, accurate, specify, and robustness with low values of LOD and LOQ. The present method met the standard criteria of ICH guidelines and can be used for analysis of amlodipine in quality control laboratories and stability indicating.

### 3.3 Forced degradation study

Forced degradation experiments provide awareness of potential degradation of the active ingredient under experimental conditions such as acid, alkaline, oxidative, thermal, and light. The present method were applied for stability-indicating study of amlodipine in two locally commercial products at various conditions as following:

**3.3.1 Acid degradation:** In stress degradation study under acidic condition, the two products of AB (Avalodipine and Amloneer) were exposed to 0.1 N HCl at 60°C up to 5 hrs. Prior to heating it was found that AB in both products started to degrade at room temperature as shown in Table 3. At 60°C temperature the degradation of both products were increased with increasing the time from 1 to 5 hrs. It can be concluded that the longer the drugs were kept under acidic and heat conditions, the more degradation can occur.

Table 3: Acid hydrolysis effect on the amlodipine in Avalodipine and Amloneer products.

sample name	Exposure conditions	Conc. Given (µg/mL)	Conc. Found (µg/mL)	Degradation %
Avalodipine	RT	10	9.72	2.78
	1h 60°C	10	8.28	17.22
	2h 60°C	10	8.00	20.00
	3h 60°C	10	7.56	24.44
	4h 60°C	10	7.28	27.22
Amloneer	5h 60°C	10	6.72	32.78
	RT	10	7.78	22.22
	1h 60°C	10	7.44	25.56
	2h 60°C	10	7.11	28.89
	3h 60°C	10	6.56	34.44
	4h 60°C	10	6.44	35.56
	5h 60°C	10	5.56	44.44

**3.3.2 Alkaline degradation:** In stress degradation study under alkaline condition, both products of AB (Avalodipine and Amloneer) were exposed to 0.1N NaOH up to 5 hrs. Preliminary, at room temperature both products were found to be unstable and lost small percentage from their actual amounts as shown in Table 4. However, when exposed to 60°C up to 5 hrs the percentage of degradation was increased with the time. The results also showed that AB in Amloneer product has less degradation than the Avalodipine product in basic and heating conditions.

**Table 4:** Alkaline hydrolysis effect on the amlodipine in Avalodipine and Amloneer products.

sample name	Exposure conditions	Conc. Given (µg/mL)	Conc. Found (µg/mL)	Degradation %
Avalodipine	RT	10	9.56	4.44
	1h 60°C	10	6.67	33.33
	2h 60°C	10	6.44	35.56
	3h 60°C	10	5.94	40.56
	4h 60°C	10	5.72	42.78
	5h 60°C	10	5.61	43.89
Amloneer	RT	10	9.67	3.33
	1h 60°C	10	8.17	18.33
	2h 60°C	10	7.00	30.00
	3h 60°C	10	6.50	35.00
	4h 60°C	10	6.39	36.11
	5h 60°C	10	6.22	37.78

**3.3.1 Oxidation degradation:** In this study, the results showed the highest degradation for AB under oxidative condition using 5% H<sub>2</sub>O<sub>2</sub>. The AB in both products started degradation at room temperature particularly with Amloneer (25 %). After heating at 60 °C, the percentage of degradation increased with time increasing from 1 to 5 hrs. Table 5 shows the results.

**Table 5:** Oxidative effect on the amlodipine in Avalodipine and Amloneer products.

sample name	Exposure conditions	Conc. Given (µg/mL)	Conc. Found (µg/mL)	Degradation %
Avalodipine	RT	10	9.17	8.33
	1h 60°C	10	7.33	26.67
	2h 60°C	10	6.33	36.67
	3h 60°C	10	3.17	68.33
	4h 60°C	10	1.22	87.78
	5h 60°C	10	0.28	97.22
Amloneer	RT	10	7.50	25.00
	1h 60°C	10	6.72	32.78
	2h 60°C	10	6.67	33.33
	3h 60°C	10	6.50	35.00
	4h 60°C	10	5.17	48.33
	5h 60°C	10	1.67	83.33

**3.3.2 Photolytic degradation:** In stress degradation study under photolytic conditions, amlodipine (Avalodipine and Amloneer) were exposed to sun, UV and dark at different day intervals. Both the Avalodipine and Amloneer products show a little degradation (less than 15%) up to 6 days and therefore the AB was found to be stable under photolytic conditions according to the ICH guideline that recommended below 15% is not measured as degradation because some of percent of this degradation is related to the drug formulation and preparation. Table 6 shows the results.

**3.3.3 Thermal degradation:** The thermal degradation was investigated exposing amlodipine (Avalodipine and Amloneer) at solid state to 60 °C for 6 hrs in an oven. The loss percentages were found to be close 11.67% and 11.1% of both Avalodipine and Amloneer, respectively. These results indicated that AB remains stable in both products according to the ICH guideline.

**Table 6:** Photolytic effect on the amlodipine in Avalodipine products.

sample name	Exposure conditions	time	Conc. Given (µg/mL)	Conc. Found (µg/mL)	Degradation%
Avalodipine	UV	1 day	10	9.89	1.11
		2 day	10	9.83	1.67
		3 day	10	9.28	7.22
		4 day	10	9.22	7.78
		5 day	10	8.78	12.22
		6 day	10	8.61	13.89
	SUN	1 day	10	9.89	1.11
		2 day	10	9.56	4.44
		3 day	10	9.39	6.11
		4 day	10	8.89	11.11
		5 day	10	7.94	13.33
		6 day	10	7.72	13.89
	DARK	1 day	10	9.89	1.11
		2 day	10	9.83	1.67
		3 day	10	9.56	4.44
		4 day	10	9.33	6.67
		5 day	10	9.22	7.78
		6 day	10	8.89	11.11
Amloneer	UV	1 day	10	9.94	0.56
		2 day	10	9.78	2.22
		3 day	10	9.67	3.33
		4 day	10	9.33	6.67
		5 day	10	9.22	7.78
		6 day	10	8.61	13.89
	SUN	1 day	10	9.94	0.56
		2 day	10	9.89	1.11
		3 day	10	9.72	2.78
		4 day	10	9.22	7.78

DARK	5 day	10	9.06	9.44
	6 day	10	8.67	13.33
	1 day	10	9.94	0.56
	2 day	10	9.72	2.78
	3 day	10	9.61	3.89
	4 day	10	9.44	5.56
	5 day	10	9.39	6.11
	6 day	10	9.33	6.67

## CONCLUSION

Nanodrop spectrophotometer method was developed and validated for determination of amlodipine in tablets dosage formulation. The develop method was found to be rapid, linear, precise, accurate, specified, sensitive, robustness' and, economic. The present method was applied successfully for forced degradation study of amlodipine of two products (Awalodipine and Amloneer) by exposing to extreme conditions including acidic, basic, oxidative, photolytic and thermal degradation. According to ICH guideline of drugs, the AB (Awalodipine and Amloneer) was found to be unstable in acidic, basic, and oxidative conditions under heating at 60°C showing high degradation percentage more than 15%. The results showed highest degradation for AB under oxidative condition using 5% H<sub>2</sub>O<sub>2</sub> with both products. The main factor that affects the degradation of AB is the passage of time. The percentage of degradation increases through different stress factors, but the effect of time remains stronger.

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